

BRIEF COMMUNICATIONS

ALKYLATION OF AMINOTHIAZOLES

V. Alkylation of 2-Ethylamino-4-Methyl-Thiazole*

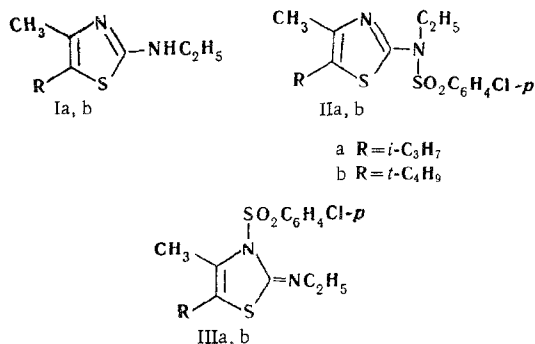
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Alkylation of 2-ethylamino-4-methylthiazole with isopropanol and tert-butanol in 85% H₂SO₄ gave 2-ethylamino-4-methyl-5-alkylthiazoles. The structure of the resultant compounds is shown by lack of diazo coupling and formation of p-chlorobenzenesulfonyl derivatives.

It was previously shown [1] that 2-methylamino-4-methylthiazoles readily undergo alkylation by secondary and tertiary alcohols to give 2-methylamino-4-methyl-5-alkylthiazoles. To ascertain to what extent alkylation of 2-alkylaminothiazoles at position 5 in the thiazole ring is general, we carried out alkylation of 2-ethylamino-4-methylthiazole with isopropanol and tert-butanol. It was found also in this case that one alkyl group was introduced to form type I compounds. The alkylation products did not undergo diazo coupling with p-nitrophenyldiazonium compound (see [2]), while treatment with p-chlorobenzenesulfonyl chloride gave acyl derivatives. Hence it can be assumed that here too the alkyl group enters at position 5 in the thiazole ring, giving 2-ethylamino-4-methyl-5-alkylthiazole (I). Obviously the acyl derivatives are 2-(N-p-chlorobenzenesulfonyl)ethylamino-4-methyl-5-alkylthiazoles (II). Formation of the isomeric 2-ethylimino-3-p-chlorobenzenesulfonyl-4-methyl-5-alkylthiazolines (III) is less probable, but we intend to make a special study of the problem.



EXPERIMENTAL

2-Ethylamino-4-methylthiazole. This was prepared from 1-ethylthiourea and bromoacetone similar to the synthesis of 2-amino-4-methylthiazole [3]. After vacuum-distillation, bp 107-110° (3-4 mm), it was obtained as an oil which crystallized on standing; colorless

prisms mp 52-53° (ex n-octane). It couples with a p-nitrophenyldiazonium salt to give a brick-red color (in pyridine vapor), on moistening with a 1 M solution of alkali, the color changed to blue. Found: N 19.69%. Calculated for C₆H₁₀N₂S. N 19.69%. The picrate was prepared by precipitating the base with ethanol with picric acid. Minute yellow needles, mp 178° (ex AcOH). Found: N 18.74%. Calculated for C₆H₁₀N₂S · C₆H₃N₃O₇. N 18.85%.

2-(N-p-chlorobenzenesulfonyl)ethylamino-4-methylthiazole. A mixture of 1.4 g (0.01 mole) 2-ethylamino-4-methylthiazole, 2.2 g (0.01 mole) p-chlorobenzene sulfonyl chloride, 1 g Na₂CO₃, and 15 ml acetone was refluxed for 1 hr, poured into water, the solid precipitate filtered off, and recrystallized from EtOH, mp 79-80°. Found: N 8.78%. Calculated for C₁₂H₁₃ClN₂O₂S₂.

Standard method of alkylation. 5.68 g (0.04 mole) 2-ethylamino-4-methylthiazole was dissolved in 70 ml 85% H₂SO₄, and at 20° for tert-butanol and 80° for isopropanol, the anhydrous alcohol (0.08 mole) added dropwise. The reaction mixture was held at the stated temperature for 10 hr (4 hr for isopropanol), cooled, poured onto ice and neutralized with conc. NH₄OH, using ice cooling. The solid alkylation product was filtered off and dried.

2-Ethylamino-4-methyl-5-tert-butylthiazole (Ib). Yield 8 g (94%). After vacuum-distillation it had bp 125-127° (3-4 mm), yield 6.4 g (80%). Coarse colorless transparent plates, mp N 14.13% (ex n-octane). Found: N 14.53%. Calculated for C₁₀H₁₈N₂S. N 14.13%. Picrate, yellow leaflets mp 181-182° (ex AcOH). Found: N 16.41%. Calculated for C₁₀H₁₈N₂S · C₆H₃N₃O₇. N 16.39%.

2-(N-p-chlorobenzenesulfonyl)ethylamino-4-methyl-5-tert-butylthiazole (IIb). Prepared similarly to the unsubstituted analog. Colorless crystals, of irregular shape, mp 82.5°. Found: N 7.81%. Calculated for C₁₆H₂₁ClN₂O₂S₂. N 7.52%.

2-Ethylamino-4-methyl-5-isopropylthiazole (Ia). Yield 7 g (95%). Bp 113-115° (3-4 mm). After vacuum-distilling yield 84%. Colorless crystals of irregular shape, mp 83-84° (ex n-heptane). Found: N 15.60%. Calculated for C₉H₁₆N₂S. N 15.20%. Picrate, yellow leaflets mp 171-172° (ex AcOH). Found: N 16.71%. Calculated for C₉H₁₆N₂S · C₆H₃N₃O₇. N 16.94%.

2-(N-p-chlorobenzenesulfonyl)ethylamino-4-methyl-5-isopropylthiazole (IIa). Pale-brown needles mp 94-95° (ex EtOH). Found: N 7.98%. Calculated for C₁₅H₁₉ClN₂O₂S₂. N 7.81%.

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*For Part IV see [1].